

GENERATION IV: Technological developments for Sodium, Gas, Lead, molten salt cooled reactors.

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Content

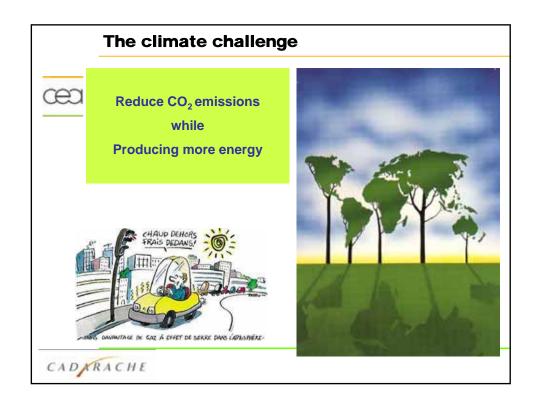


French Nuclear policy
Generation IV concepts
Sodium cooled Fast Neutron reactors
Gas cooled Fast Neutron reactors
Other developments (Lead alloys, salts)

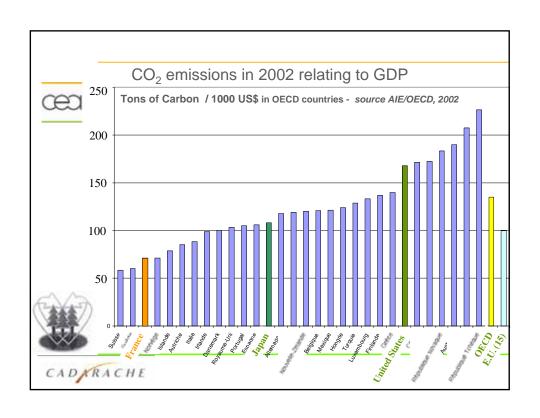
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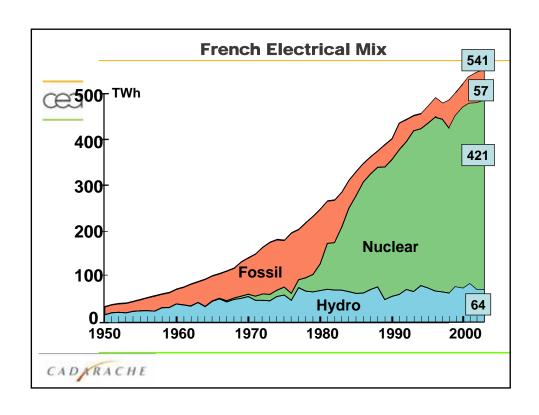


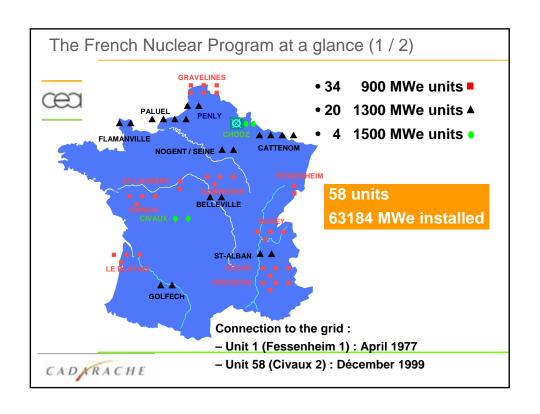
French Nuclear policy

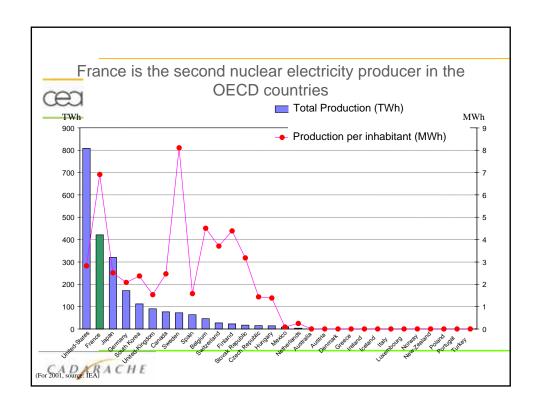


10.5		
CO2 emission	per kWh	Direct emission p.a (for a 1000 MW*
Nuclear	2-6 g	O Mt
Gas	400 g	3 Mt
Coal	800-1000 g	6-7 Mt
	15 S	* for 7.5 TWh/year (KP = 85,6%)









Today: Main features of the French energy status



Energy import cost of 22,7 billions Euros in 2003, accounting for 1,4 % of French GDP (versus 5 % in the 80's);

The rate of energy independence has considerably increased, from 26 % in 1973 to 50 % today.

The fourth highest energy consumer in the OECD countries, but only ranked 27th for CO₂ emissions (2002, AIE).

A competitive electricity supply for companies and households, characterised by steady prices.

A self-sufficient national electricity production allowing France to be the first electricity exporter in the world.

The French nuclear policy: the main actors



Definition of the French nuclear policy : DGEMP (Direction Générale de l'Énergie et des Matières Premières, General Directorate for Energy and Raw Materials)

Research and development in the nuclear field: CEA (Commissariat à l'Énergie Atomique, Atomic Energy Commissariat), IRSN (Institut de Radioprotection et de Sûreté Nucléaire, Institute for Radiological Protection and Nuclear Safety)

Companies : AREVA (Framatome-ANP, COGEMA), Alstom

Utilities: EDF

Waste management : ANDRA (Agence Nationale pour la gestion des Déchets Radioactifs, National Agency for Radioactive Waste Management)

Safety control: DGSNR (Directorate General for Nuclear Safety and Radiation protection) (expert: IRSN)

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The French National Debate on Energy



2003 : « National Debate on Energy », launched by the Ministry of Industry



2004: First draft of the Energy Orientation Bill, adopted by the Council of Ministers on May 2004 and adopted on second reading by the National Assembly on the 29th of March 2005 and will be put to the Senate in May 2005

Reaffirms the necessity of nuclear power, authorizes EDF to build a first EPR and encourages R&D for the future nuclear systems.

Scenario for the renewal of French power reactors based on Gen III reactors (EPR) deployment with a transition to Gen IV fast neutrons systems as from 2040

The 3 challenges structuring French energy policy

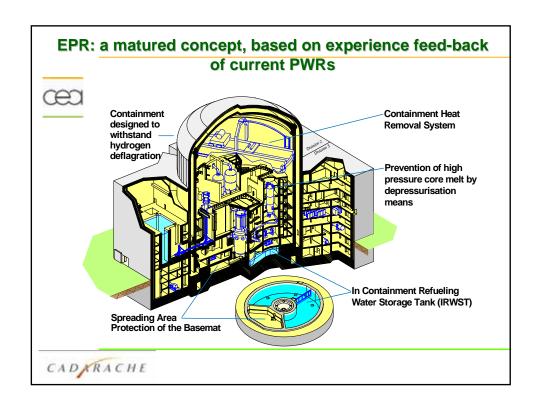


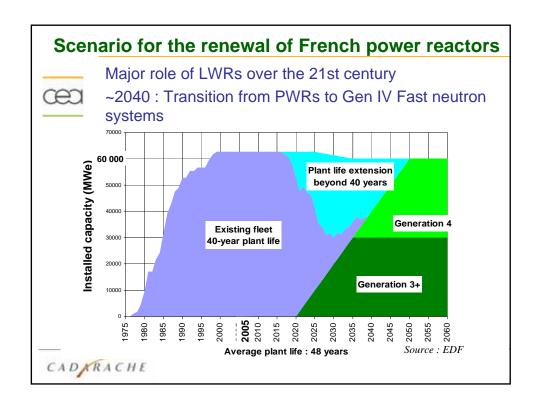
First challenge: reducing our green house contribution by dividing by 4 our green house gases emissions by 2050

Second challenge: answering the gas and oil prices growing due to the increase of the global demand in the framework of ressources run out before the end of the century

Third challenge: facing the anticipated renewal of our nuclear fleet

- <u>2015 2020</u>: beginning of current fleet renewal building based on evolved and mature PWR (EPR).
- beyond (> 2030 2040 ?), completing the renewal in using Generation IV reactors concept.





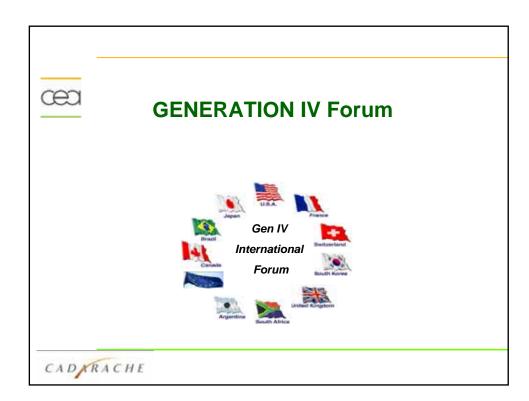
R&D Strategy of France for Future Nuclear Energy Systems



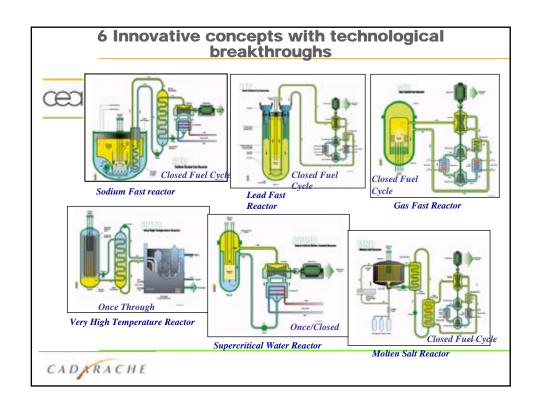
Approved by the French Government in March 2005

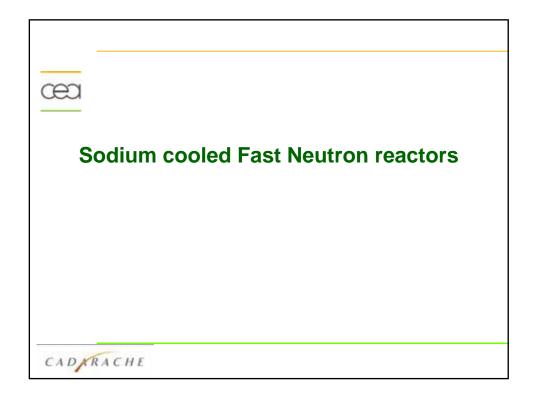
- 1 Development of Fast Reactors with closed fuel cycles, along 2 tracks:
 - Sodium Fast Reactor (SFR)
 - Gas Fast Reactor (GFR)
 - > New processes for spent fuel treatment and recycling
 - → Industrial deployment around 2040
- 2 –Hydrogen production and very high temperature process heat supply to the industry
 - Very High Temperature Reactor (VHTR)
 - Water splitting processes
- 3 Innovations for LWRs (Fuel, Systems...)

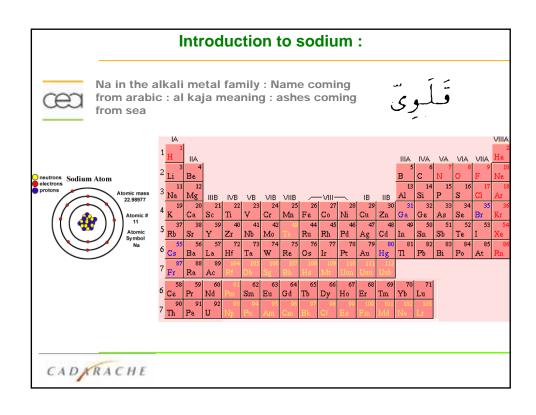


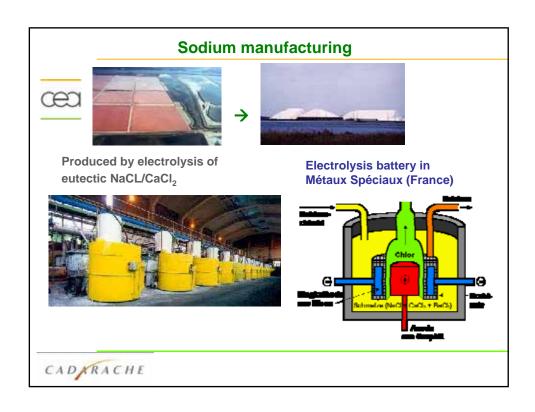


Ensure energy needs are met in the long term without emitting greenhouse gases > Gradual improvements > Economic competitiveness > Safety and reliability > Significant steps forward: > Saving of natural resources > Waste minimization > Security: non-proliferation, physical protection > An opening to other applications: > High temperature heat for industry > Hydrogen vector > Drinking water









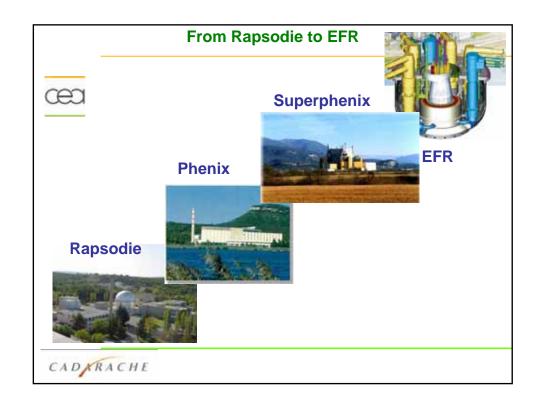
Sodium properties

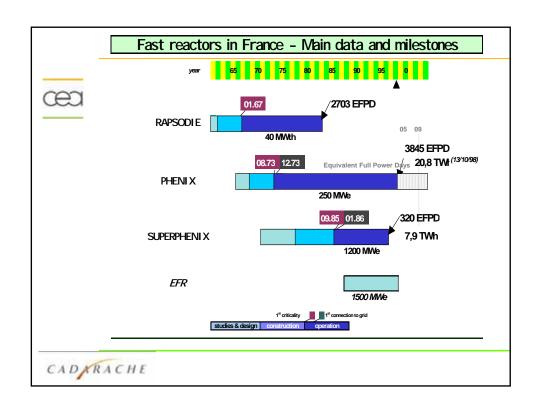


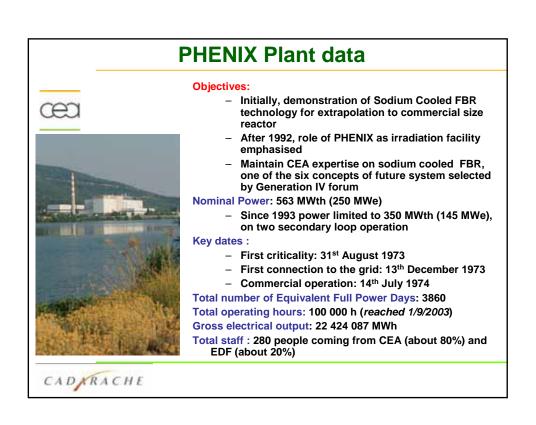
Na a very attractive coolant for Fast Neutron Reactors :

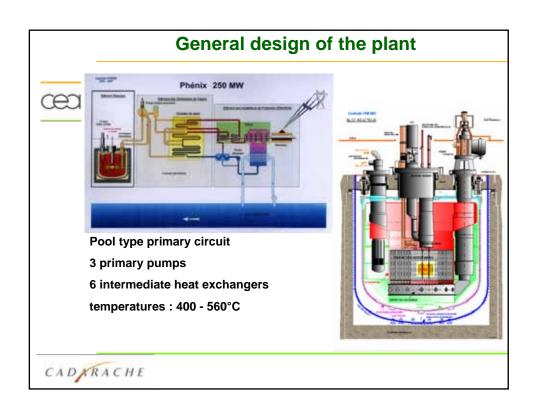
- Very good thermal conductivity.
- High thermal capacity.
- Liquid between 97.8 up to 880°C at dynamic pressure below 4 bars,
- Compatible neutron-physical properties.
- Viscosity comparable to that of water.
- Compatibility with metallic materials fairly satisfactory.
- No toxicity
- Low cost,...

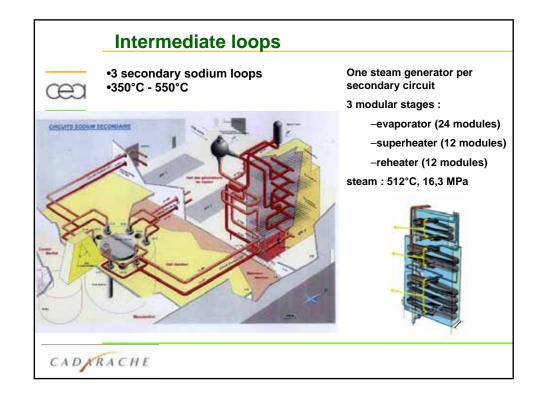
But high reactivity











Some incidents



sodium leaks

- ☐ from a few grams to twenty kilograms
- most of the leaks located of on welds of secondary loops and auxiliary circuits
- leak satisfactorily detected in an early stage (significant corrosion due to delayed detection on one occasion: lead to improvement of detection system)

5 small sodium-water reactions (4 in 1982 - 1983 and one in 2003)

- ☐ Efficiency of the hydrogen detection
- □ replacement of the modules
- Modification of the procedures

4 negative reactivity trips in 1989-1990

- Extensive studies and testing to identify the cause of the transient
- International Expert committee : 200 Man. Year 500 documents
- Examination of all possible phenomena
- Safety analysis conclusions: no safety consequences for the reactor



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LIFETIME EXTENSION PROGRAMME



Core and primary circuit Safety up-grading

- Installation of one safety control rod
- SARA fast acquisition system
 - Instrument devices in the hot and cold pools

Seismic reinforcement of steam generator building

Protection of the SG Building against

sodium fires (Resistance to the reference earthquake, sodium fire of 1100°C degrees for 30 ')

Re-evaluation of damages to the block structures

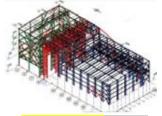
Major component repairs and replacements :

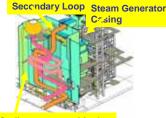
Repair of 321 steel components subject to delayed reheat cracking

- secondary loop hot legs
- SGU sodium headers
- SGU modules

Manufacturing of three new IHX

Primary and secondary pumps refurbishing,....



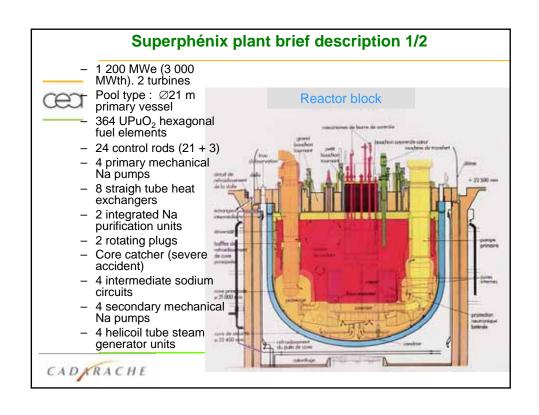


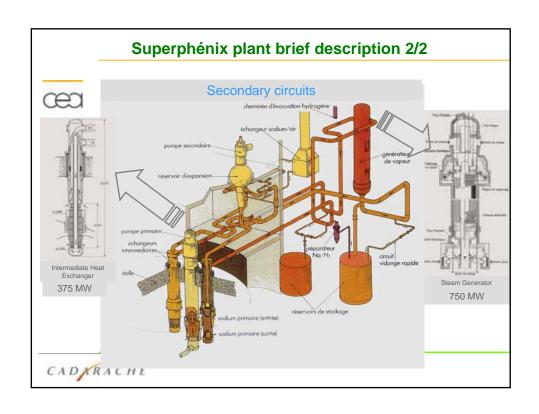
Sodium zone partitioning

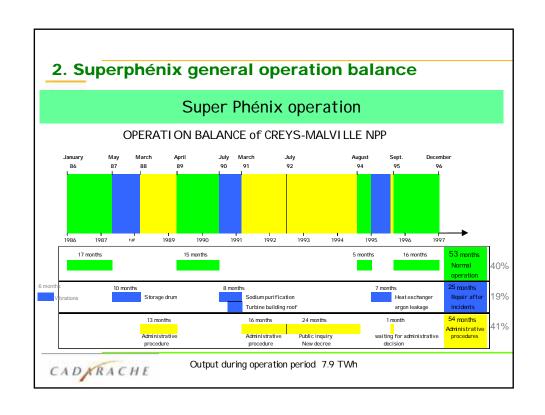
4 years of work and of requalification.

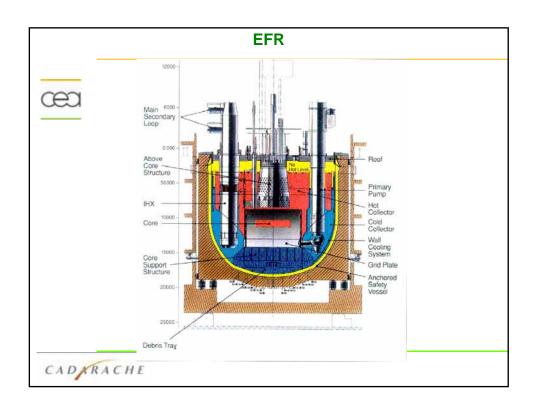
160 million € of studies and renovation works.4000 contractors from 350 companies











Main objectives for the Sodium Fast Reactors 1/2



- improvement of safety:

- severe accidents have to be mastered: potential initiators of accidents and subsequent energy release. Mainly: Core safety and notably problems associated with criticality control (void-effect, recriticality)
- specific risks due to Na must to be minimized (or avoided): Na fire, Na-water interaction.
- improvement of competitiveness in order to maintain the attractivity with regards to the existing operated reactors.
 - Reduce cost of investment by design simplification and improvement of efficiency: energy conversion, burn-up, systems&components reliability and reactor availability,...)
 - Improvement of In-service inspection and repair,
 - Life extension up to 60 years.
- advanced materials for structures and fuel
- saving of natural ressources : U

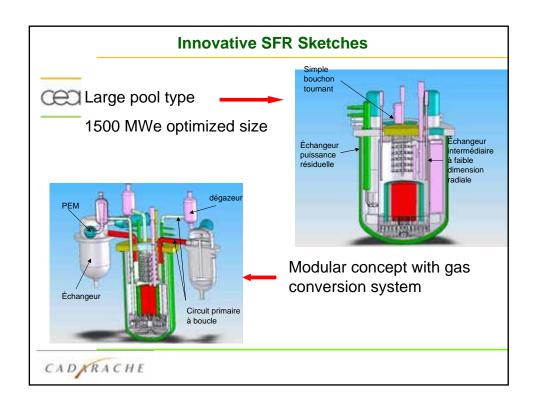


Main objectives for the Sodium Fast Reactors 2/2



- non proliferation: systems which regenerates plutonium without lateral fertile fuel assemblies.
- waste management by transmutation and waste minimization (for irradiated fuel reprocessing and reactor operation and maintenance)
- minimization of environmental impact : liquid and gaseous effluents, operators dosimetry, decommissioning strategy,...
- definition of 3 to 4 concepts for a sodium cooled reactor :
 - to illustrate proposed innovations within a global design,
 - to evaluate resulting economics and associated risks,
 - to best target the most promissing R&D paths,
 - to contribute to making a project come true.



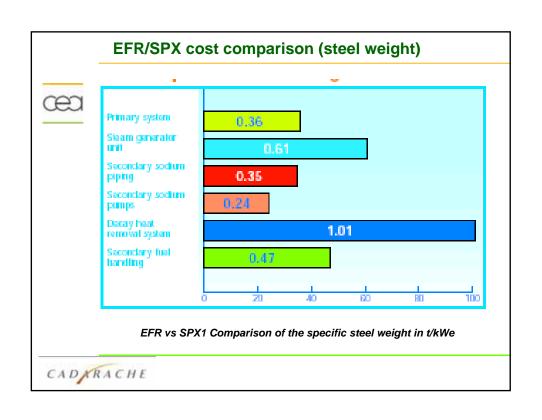


Improvement of safety:



- Reactor designed in order to reduce strongly the probability of core melting; nevertheless a reference accident involving a core melting event will be postulated, investigated and demonstrated as mastered.
- Protection against plane shut-down,...
- Elimination of environmental consequences outside the nuclear plant; necessity to focus on incidents involving Na, (sodium leaks, release):
 - Sodium pipes and/or systems inertization
 - Sodium leak control systems : reliability improvement
 - Optimized In Service Inspection strategy
 - Efficient, safe and fast Repair operations.
 - Minimization of Liquid&gaseous effluent releases (« zero » release?),...

ea	On site workshop	2,7
	Reactor assembly structures	19,4
	secondary handling	5,7
	secondary circuit	9,0
	DRACS	3,7
	SGU	11,0
	primary auxiliary system	48,6
		100,0



Improvement of competitiveness



Reduced cost by:

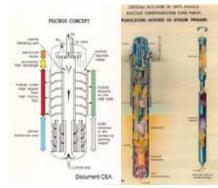
- design simplification :
 - intermediate loops : simplification, shortening or suppression,
- ancillary systems (for cover gas control, sodium purification, measurements,...): minimization of the number of circuits and/or simplification, after reassessment of functional specifications and operation feedback analysis.
- fuel assemblies handling systems, after reassessment of functional specifications and operation feedback analysis.
- improvement of efficiency:
 - energy conversion
 - high burn-up,
- reactor availability increased by : systems&components reliability
 - improvement of systems&components reliability
 - reduction of fuel assemblies handling duration,...
- extension of life duration : up to 60 years.



Ancillary systems : ex : Purification systems



- Cost reduction is possible, thanks to :
 - reassessment of functional specifications
 - feedback from reactors in operation and design studies (EFR,...)
- Purification systems:
- primary vessel:
 - SPX: 2 integrated cold traps
 - EFR: 2 external cold traps
- Intermédiate loops : 1 purification loop for each intermediate loop
- Ancillary Na systems : SPX and EFR : purification loop for each intermediate loop





Primary purification system:

Requirement for [0]:

œ

Two main options:

[O] < 3 ppm (<5ppm during 1 month) (strategy of « clean » Na) or

[O] < X ppm (X>3 : as exemple 10 ppm) (UK)

Consequences:

- improvement of tribological behaviour but :
- increase of dosimetry (due to activated corrosion products), increased radionuclides inventory in cold traps, increase of liquid effluents (due to cleaning/decontamination).

Corrosion rate depends on mainly [O], primary sodium and fuel pin temperatures, sodium velocity

- → necessity to compare PX and PFR feedback.
- → necessity to model oxygen behaviour and compare corrosion and purification kinetics: effect of reactor size.

→ options for purification systems :

- → continuous or non-continuous operation. Impact of : specification on [O], incidental pollution, tritium balance of the reactor, defueling?)
- → cold trap always in situ or removable
- cold trap with cartridge or not (makes cold trap treatment easiest or not, and consequences on the size of the cold trap and building allocated to this system.
- > Other specific concepts



Intermediate purification system

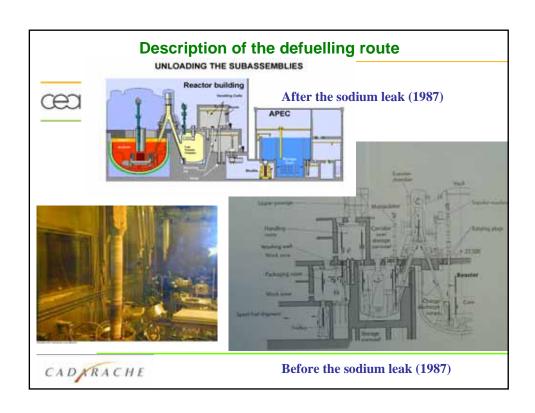
Case: Rankine Cycle (H₂O):



Specifications on:

- nominal residual [H] : with regards to efficient fast water-sodium detection in SGU
- tritium release (EFR : difficulty to fulfill requirements due to choice of ferritic steel Steam Generator : aqueous corrosion increased→ continuous H source increased)
- → H trapping in intermediate cold traps allows to trap also tritium and to control release.
- incidental purification campaign : anticipated Na2O and NaH inventory.
- → options for purification systems :
- → continuous or non-continuous operation ? Impact of : specification on [H], incidental pollution, tritium balance of the reactor,)
- → cold trap always in situ or removable
- → cold trap able to be regenerated or not, with cartridge or not (makes cold trap treatment easiest or not, and consequences on the size of the cold trap and building allocated to this system.
- → cold trap for continuous and incidental pollution or not (specific removable cold trap for incidental pollution, able to be installed on the involved loop)
- → other concepts?





Technological dysfunctions: first defuelling operation as routine procedure, 15 years after SPX start-up. (delay sufficient to raise the probability of dysfunctions of instrumentation) Unsuitability with the safety requirements: The first safety document was limiting the total residual amount of sodium to 300 grams. Assessment too restrictive: The cleaning pit has demonstrated it could treat greater amounts of sodium without any risk. This assessment had major consequences: loss of 16 weeks in the defuelling planning, modification of the future cleaning procedure that decreases the potential defuelling rate. Average defuelling rate: 4.5 per week (Expected defuelling rate: 9 assemblies/week) → The defuelling procedure was not as efficient as expected because cleaning processes were too long and defuelling route for one assembly too complex. Absence of an EVST appears to be crucial. Incidents Unsubability with Softment 19% (Poperating time 19%) Technologic mysfunctions 19%

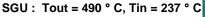
Why other energy conversion systems?

SPX:

Tout (core) = $530 \, ^{\circ} \, \text{C}$,



Tin (core) = 377 ° C



Pout = 180b

Net efficiency = 40%





Why other cycles? (SC H2O, SC CO2, N2, N2-He,...)?

Economical attractivity:

- Obtain a better thermodynamical efficiency with the same core,
- Simplify or eliminate the intermediate loops,
- Reduce the cost of the energy conversion system (i.e small turbine, ...)



Safety: better compatibility with Na than water but

potential gas injection in primary Na to be investigated and prevented.

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Design options and main consequences



Energy Conversion System (ECS) coolant : H_2O , CO_2 , N_2 ,... \rightarrow impact on :

- → ECS thermodynamic efficiency
- → heat exchanger design (efficiency, reliability, compacity, inspectability, reparability/modularity,...)
- →interaction between ECS coolant and structural material
- **→**interaction between ECS coolant and sodium <u>or</u> ECS coolant intermediate fluid (Ga, Pb-Bi,...) in case of integrated IHX-SGU component.
- → Behaviour of products from interactions : content, dissolved phase/particles,...)
- → Leak detection systems
- → purification (filters, limited effect of cold traps,...)
- → Thermal/mechanical/chemical effect of ECS coolant (westage)
- → transfer of tritium between primay circuit and ECS,...

Influence of various parameters on overall efficiency



Tinlet core

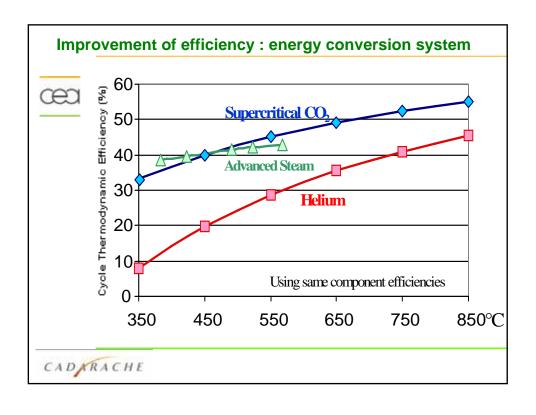
Toutlet core

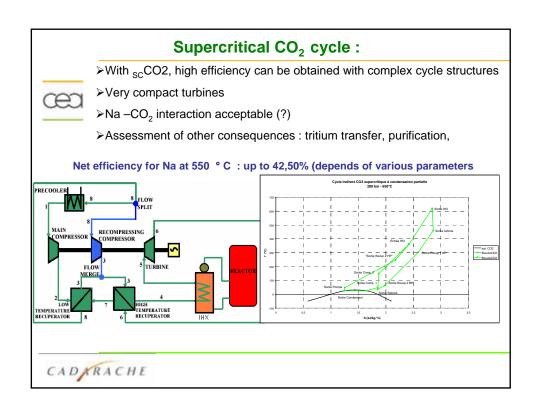
 ${\sf Gas:N_2,N_2\text{-}He,N_2\text{-}Ar,He,CO_2,}$

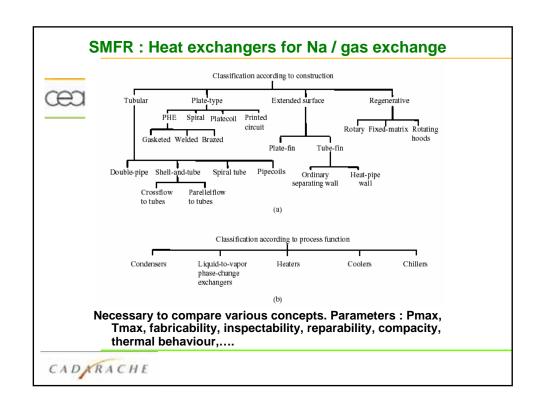
Gas pressure

Cycle structure

Efficiency of single components (compressor, heat exchanger, recuperator, turbine,...







Shell and tube heat exchangers



o A shell and tube heat exchanger is essentially a bundle of tubes enclosed in a schell and so arranged that one fluid flows through the tubes and another fluid flows across the outside of the tubes, heat being transferred from one fluid to the other through the tube wall.

o Shell and tube heat exchanger have been constructed with heat transfer areas from less than 0.1 m² to over 100 000 m²

o for **pressures** from deep vacuum to over 1000 bar .

o for temperatures from near 0 to over 1400 K

o for **all fluid** services including singlephase heating and cooling and multiphase vaporization and condensation.

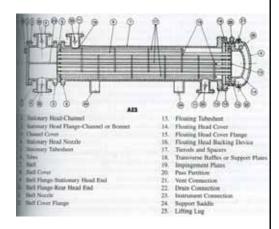




Plate-Fin Heat Exchangers

- Aluminium plate-fin heat exchangers (PFHE) were initially developed in the 40's to provide for the aerospace industry compact, light and high efficient heat exchangers for gas/gas applications.

The fins and the parting sheets are assembled by fusion of a brazing alloy cladded to the surface of the parting sheets. The brazing operation is made in a vacuum furnace in which the brazing alloy is heated to its point of fusion.

Numerous fin corrugations have been developed, each with its own special characteristics.

For higher temperature applications stainless steel (temperature up to 700°C) or copper materials can be used.

For very high temperature (gas turbine heat recovery ; T > 1200 $^{\circ}$ C), a ceramic plate fin heat exchanger has also been developed





Pictures of plate fin heat exchangers





Different fin geometry





Stainless steel brazed plate fin heat exchanger (courtsey of Nordon)



Microchannels Heat Exchangers



Microchannels heat exchangers refer to compact heat exchangers where the channel size is around or lower than 1 mm. (developed for severe environment such as offshore

platforms)

New applications are also arising for nuclear high temperature reactors (Heatric Company)

To manufacture such small channels several technologies are available: chemical etching, micromachining, electrodischarge machining ...

The processing technique is as flexible as for plate-fin heat exchangers, and crossflow and counterflow configurations are employed.

The main limitation of microchannel heat exchanger is the pressure drop, which is roughly inversely proportional to the channel diameter.





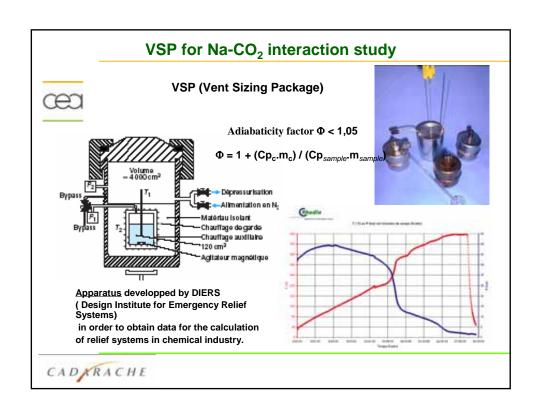
Detail of the bonded plates

Printed circuit heat exchanger (courtesy of Heatric)



View of completed unit: A typical 24.4 MW unit operating at 70 bar

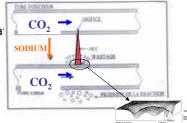
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Na-CO2 interaction

- Exothermal reaction Na + Na-oxalate and CO production then exothermicity due to Na + CO (with induction time) around 400 ° C over this temperature.

- → several by-products and complex exothermicity
- Limited reaction Na + carbonate ?
- T> 500 ° C interaction producing directly carbonate
- T< 500 ° C more complex scenario
- Study going on; (to be published in near future)
- → With the aim to check :
 - the absence of wastage,
 - the significative dissolution of carbona
 - the very limited effect of carbon,
 - The efficiency of detection systems





Comparison Na-CO₂, Na-H₂O (1/2)



Event	Water ingress	CO ₂ ingress	Comments
Pressure (H ₂ O or CO ₂)	168 bar	200 bar	Impact of pressure on heat exchanger design
ΔH (kJ/mol)	162	271.5	
Kinetics of reaction	Almost instantaneous	Order 2 / CO ₂	For CO ₂ to be investigated in realistic conditions
Products of reaction	NaOH, O ⁼ , H⁻	Na ₂ CO ₃ , C, and Na ₂ C ₂ , Na ₂ C ₂ O ₄ as intermediate products	Necessity to investigate carbonate dissolution in sodium
Particulates?	Not	Yes (Na ₂ CO ₃)	Necessity to foresee filters
Soluble products trapping	Cold trap, hot trap	To be defined	Necessity to foresee a carbon trap?
Non soluble products trapping	Not necessary	Filters,	Implementation has to deal with thermo-hydraulics



Comparison Na-CO₂, Na-H₂O (2/2)



Event	Water ingress	CO ₂ ingress	Comments
General corrosion	Yes but very limited due to NaOH conversion with Na	Not	
Carburization	No	Yes at high temperature and high C content	Probably very limited consequences
Possibility of Leak detection	Yes (hydrogen-meter, acoustic detection system,)	Yes (diffusion or electrochemical carbon meter, sampling then analysis,)	Necessity to assess the response time for CO ₂
Wastage	Yes	Probably not	To be investigated
Risk of strong reaction	Yes, if contact with oxygen	Not	Probability of loss of tightness of gas plenum to be investigated
Risk of combustion	Not	Yes	Operating conditions to be investigated

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ISIR: main objectives



In Service Inspection and Repair (ISI&R) is considered as one of the problems to be resolved for future liquid metal cooled reactors SFR.

Two main applications can be cited:

- Non Destructive Examination (NDE) of these structures
- Telemetry within liquid metal coolant (and also vision as a connected item)

What we are looking for?

Material failure

 $Loss\ of\ thickness\ (tribology,\ we stage\ effect, \ldots)$

Fatigue/ creep/ local effect of temperature

Manufacturing defaults (dye penetration,...)

Horizontality of core

Right position of removable systems

Respective positions of systems/components

Loss of Tightness

Erosion/cavitation effects

Metal embrittlement,...



Main difficulties for ISIR in Na



Na opacity

Impossibility to drain primary Na and unload core Fuel Assemblies

High activation importante of internals structural material

Temperature: 160 ° C-200 ° C, during cold shut-down

Na chemical reactivity

Necessity of confinement with regards to air and water

Tribology for holders in Na or air

Wetting characteristics (and criteria) of Na on traducers surfaces

Dose rates in Na (up to 500 Gray/h) and possible contamination (Pu...)



ISIR technologies



Optic control (endoscope, periscope, dye penetration, chemical tests,

Infra Red Camera,...

Mechanical metrology

Volumetric control by US

Under Na US telemetry or in air telemetry

Laser telemetry in gaz

Acoustic detection (rotating machines, Na-water interaction, gas leak in Na,...)

Main parameters monitoring: temperatures, Na level, flow-rates, pressure, rotating speed, neutronic parameters...

Sodium leak detection systems, water-Na interaction detection systems,

System/components tightness: He-Ar-N2 detection, pressure monitoring, activity monitoring,...

Components or sampling devices dismantling and inspection and analysis in dedicated laboratories,



Parameters impacting ISIR:

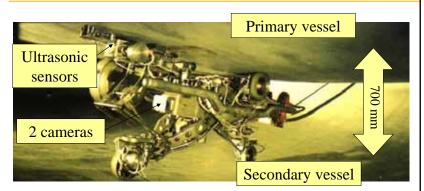


- tools and processes
- simplicity in the design
- easy access
- characteristics of structures
- size of structures
- modularity
- removability
- conditions for intervention

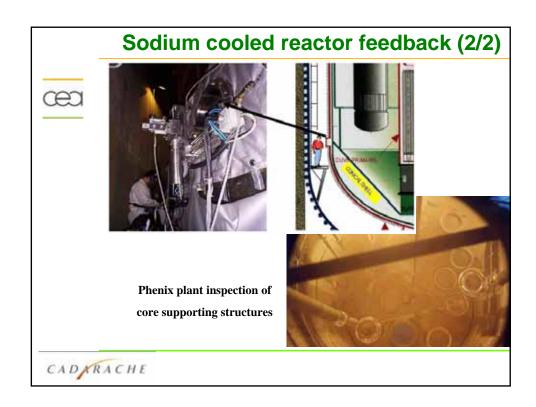
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Sodium cooled reactor feedback (1/2)



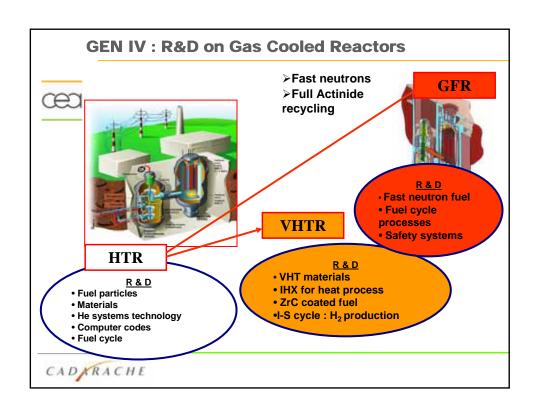


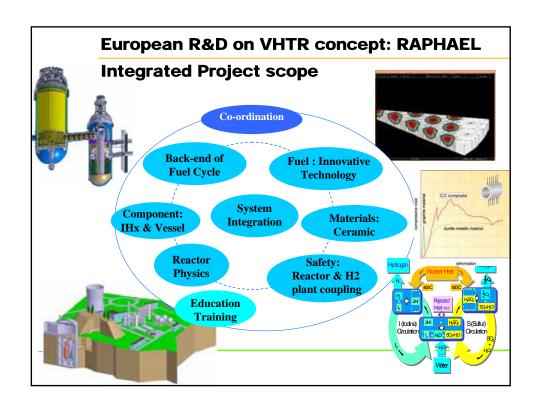
Inspection Engine
of Superphenix plant Primary Vessel





Gas cooled Fast Neutron reactors





Very High Temperature Reactor – (VHTR)



Main goals:

Calculation system

Particle fuel

Very high temperature resistant

materials (> 950°C)

High temperature helium circuit technology

Hydrogen production processes

Conversion system



Main Milestones

> 2009: Feasibility

> 2015: Confirmation of performance







Tribology

<u>Fuel</u>

Gas Fast Reactor (GFR)



Fuel and core materials

Choice of fuel reference materials (composite ceramic, ceramic cladding, etc.)

Fuel cycle reference technology

Accident management (depressurisation, etc. ..) High temperature helium circuit technology Applications technology (Gas turbine, etc...)

GFR fuel



Main Milestones

➤ 2012: Feasibility

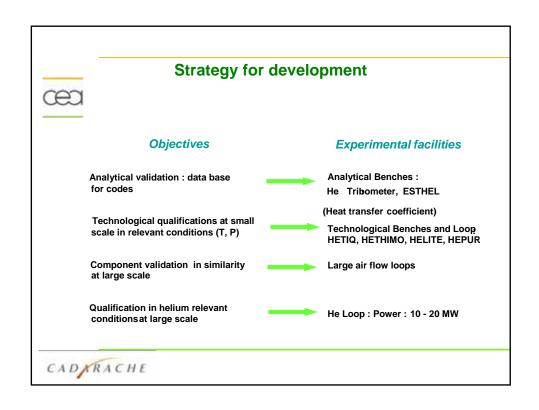
> 2020: Confirmation of performance

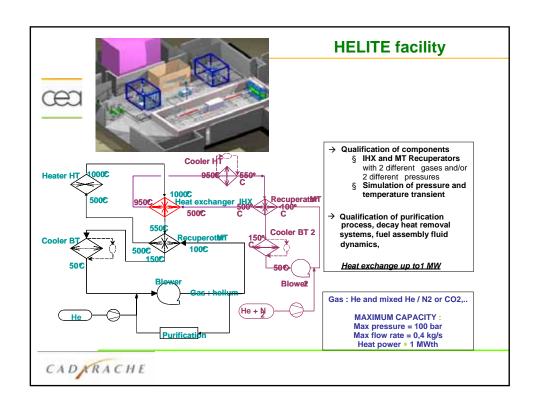
> 2025+: GFR 1st Demonstrator



1.2 GWe GFR







R&D facilities for He technology in Cadarache



HEDYS: validation of He Dynamic Seals for rotating shaft

HETAP : functional and thermo-mechanical validation of high T heating elements

HETHIMO: Thermal insulation system qualification

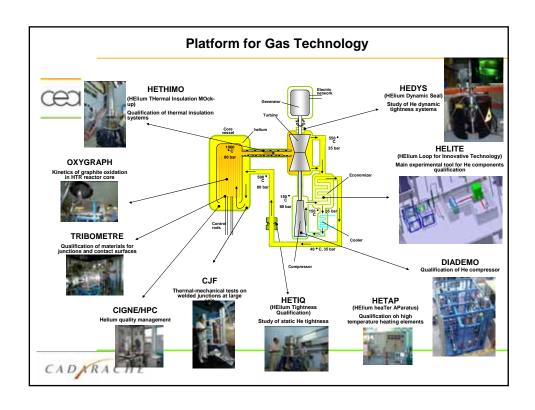
HETIQ : Validation of tightness systems; qualifications of seals OXYGRAPH : Kinetics of graphite oxidation in HTR reactor core

TRIBOMETER: Qualification of materials for junctions and contact surfaces

CIGNE/HPC : Helium quality management DIADEMO : Qualification of He compressor

HEDYT: Generic qualification bench for He technology.

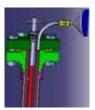




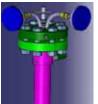
HETAP: HE heaTer APparatus



 $\underline{\textit{CONTEXT}:} \ \text{In the frame of the He technology development program for GCRs, the HETAP}$ experimental facility is dedicated to the functional and thermo-mechanical validation of high temperature heating elements



Heating



Holes for

instrumentation

PARAMETERS: Atmosphere: helium,

Pressure: up to 100 bar,

He flow rate : up to 300 NI/min, Temperature: up to 1000°C.

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EXPERIMENTAL PROGRAM

Mechanical behavior of the sheath:

- Tests at 75% of full power,
- ■T sheath = 1050°C,
- P=100 bar,
- No He flow
- Thermal cycling and endurance tests

Limits of performance :

- Tests at full power,
- P=100 bar,
- ■He flow up to 1000°C
- Thermal cycling and endurance tests

HETHIMO: HE THermal Insulation MOck up 1/2

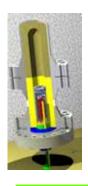
CONTEXT: Gas Cooled Reactors (GCR) use high pressure (100 bar) and high temperature (up to 1000°C) gas. In fact, no design code predicts this loading level with listed materials.



It is necessary to thermally insulate and/or to cool down the structural material in order to withstand such pressure load.

OBJECTIVE: Thermal insulation system qualification for GCRs

- thermal performance,
- mechanical behavior in case of high depressurization rate.







Thermal insulation mock-up

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HETHIMO: HE THermal Insulation MOck up 2/2









Thermal insulation mock-up

Experimental program:

- Endurance test,
- Thermal cycling,
- High depressurisation rate

Operating conditions:

- P = 100 bar,
- T = 1000 ° C,
- ΔP/ Δt = 20 bar/s
- He and He/N2



HETIQ: HE Tightness Qualification 1/2



CONTEXT: Gas Cooled Reactors (GCR) use helium (up to 1000°C and 100 bar) as coolant. Due to its small molecular size, the helium gas easily leaks.



For GCRs, it is crucial to confine He circuit in order to :

- · limit operation cost,
- reduce contamination level.

OBJECTIVE: Seals qualification in GCRs conditions (100 bar, 1000 °C, thermal and pressure cycling)

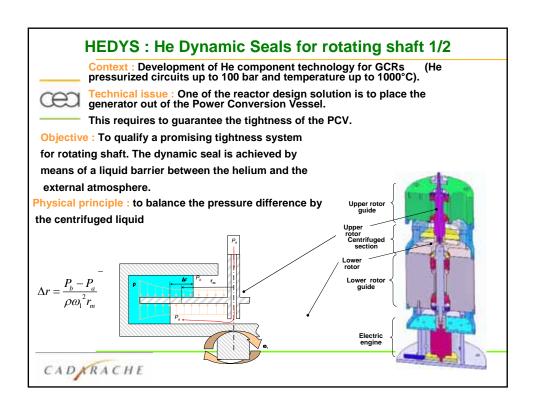
Experimental measurements:

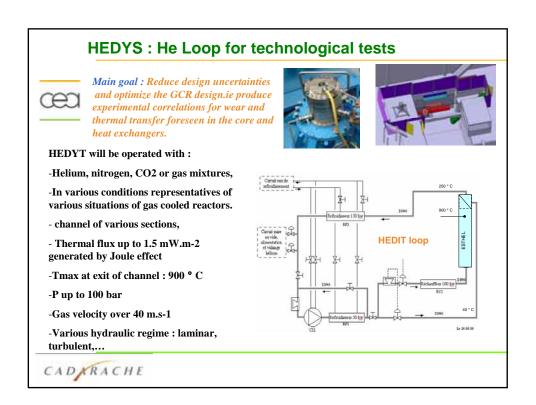
- •He pressure and temperature,
- Tensile stress on bolts
- •He leak flow rate with mass spectrometer

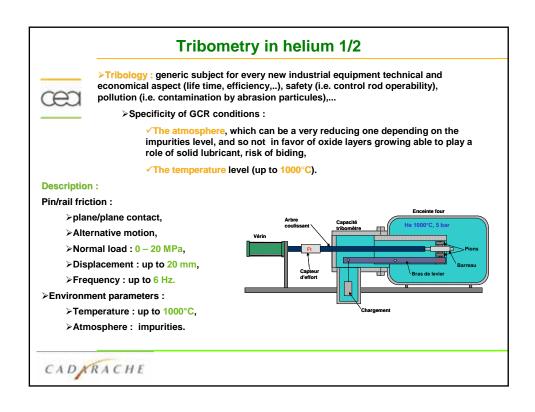


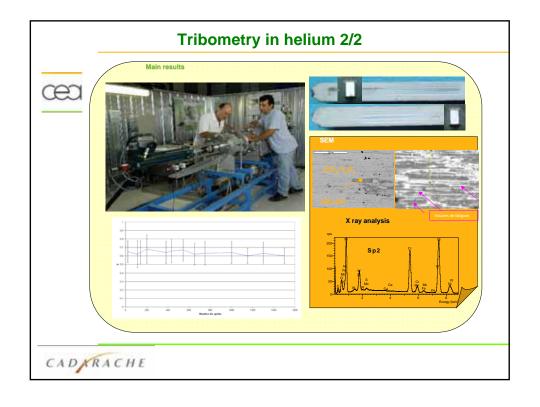












He quality control and purification 2/3



Necessity to maintain impurities concentration at low levels and under some thresholds [1];

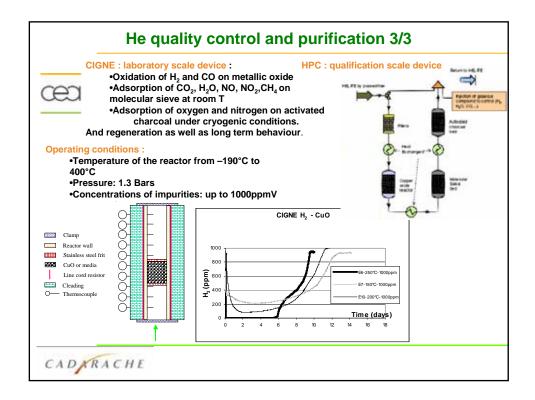
Determination of values from litterature and synthesis of constraints (material corrosion, maximum impurities concentration when circulating and eligible contamination threshold ...):

- → 1 For instance, concerning the Peach Bottom reactor and for a working pressure of 24 bar, concentration levels are:
- o CO < 10 ppm V
- o CO₂ < 2 ppm V
- o H₂ < 10 ppm V
- o H₂O < 0.5 ppm V
- o CH₄ < 2 ppm V
 - 2 Resistance to corrosion. For example, to set an oxide film for protection of these type of material (supposing a chromium activity of 0.3) following conditions should be applied:

- at 950°C: a_{O2} > 2.10⁻²³ μbar; i.e. P_{H2O}/P_{H2}>10⁻⁴ and P_{CO}> ~150μbar at 1000°C: a_{O2} >10⁻²¹ μbar; i.e. P_{H2O}/P_{H2}>10⁻⁴ and P_{CO}> ~500μbar and necessity to adress also: graphite oxidation, fission products (poisoning due to Xe, radiological inventory in case of accident, dosimetry,...) and graphite aerosols (erosion, clogging,...).

[1], Helium purification at laboratory scale Conf. HTR 2006 Fanny Legros and all





OXYGRAPH: High Temperatures Graphite Oxidation

General objectives :

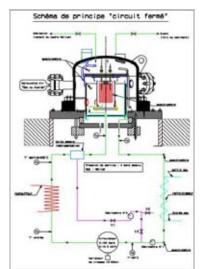


To simulate in realistic conditions an air ingress on CGR graphite components.

> -To supply and complete the existing data bases to validate the oxidation models and the related calculations.

Experimental goals:

- Check the behavior of the selected nuclear graphite grades under oxidizing conditions.
- Define oxidation kinetics parameters f(T, %W
- Effect of the Oxidation on mechanical characteristics (compressive strength loss).
- Consequences from the safety point of view (carbon monoxide production, release of radioactive products formerly trapped in graphite...)
- Two configurations : Open loop (simulation of a massive and continuous air ingress) and closed loop (simulation of a limited air ingress)

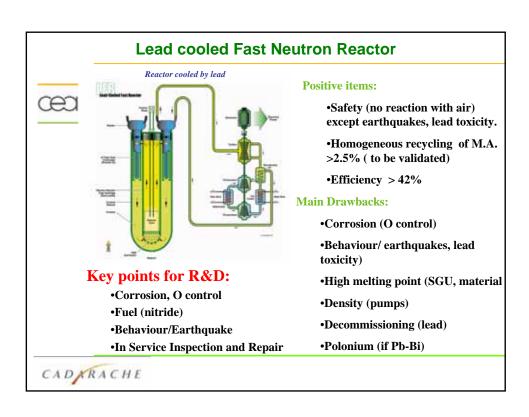






Lead cooled Fast Neutron Reactor





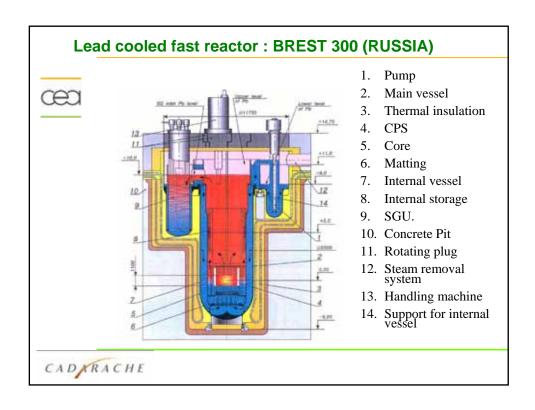
Some Heavy liquid metal properties:

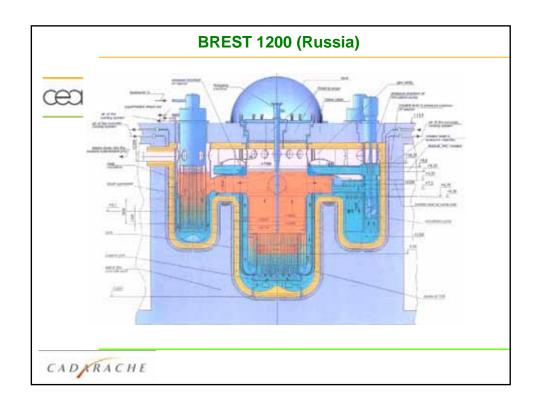


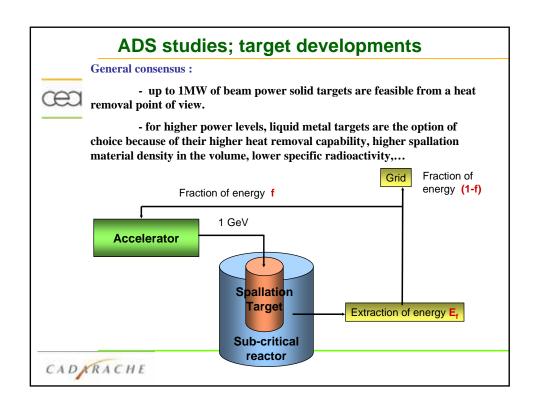
Property		Pb	Bi	LME *	LBE**	Hg
Composition		elem.	elem.	Pb	Pb	elem.
				97.5%	45% Bi	
				Mg 2.5%	55%	
Atomic mass A (g/mole)		207.2	209	202.6	208.2	200.6
Density	20°C	11.35	9.75			10.5
(g/cm³)	liquid	10.7	10.07	10.6	10.5	13.55
Linear coefficient of	solid	2.91	1.75			
thermal expansion (10 ⁻⁵ K ⁻¹)	liquid (400°C)	4		4		6.1
Volume change upon		3.32	-3.35	3.3	0	
solidification (%)						
Melting point (°C)		327.5	271.3	250	125	-38.87
Boiling point at 1 atm (°C)		1740	1560			356.58
Specific heat (J/gK)		0.14	0.15	0.15	0.15	0.12
Thermal neutron absorption (barn)		0.17	0.034	0.17	0.11	389

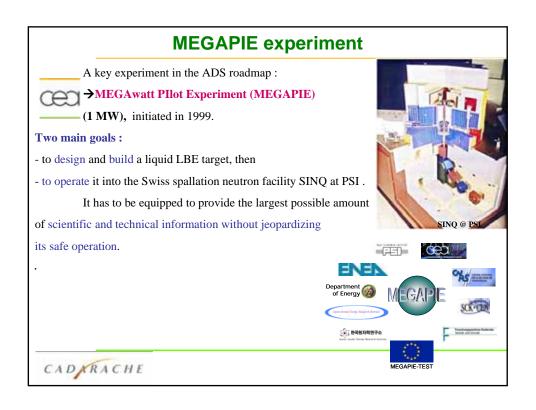
LME - lead/magnesium eutectic ** LBE - lead/bismuth eutectic

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Choice of Lead-Bismuth



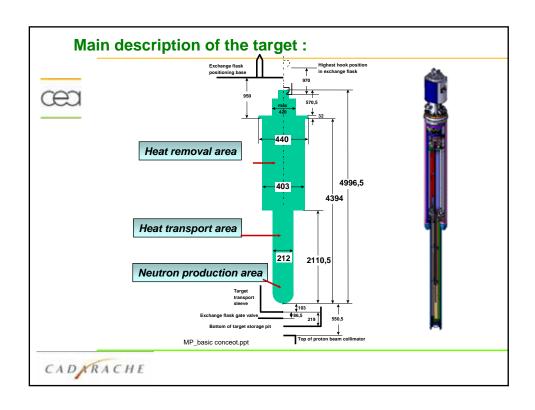
Lead bismuth eutectic (Pb44.5%-Bi55.5%) has been selected, due to its attractive neutronic and physical properties : heat transfer coefficient, low melting point (125 $^{\circ}$ C);

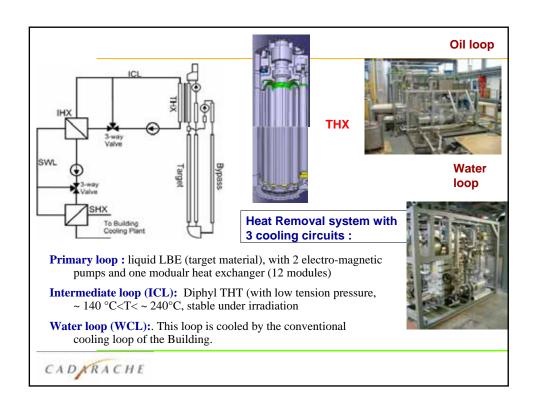
nevertheless bismuth induces to the production of activation products i.e. polonium,...

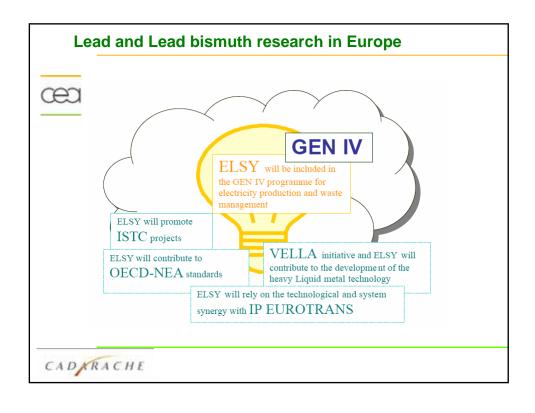
Property		Pb	Bi	LME *	LBE**	Hg
Composition		elem.	elem.	Pb	Pb	elem.
				97.5%	45% Bi	
				Mg 2.5%	55%	
Atomic mass A (g/mole)		207.2	209	202.6	208.2	200.6
Density	20°C	11.35	9.75			10.5
(g/cm ³)	liquid	10.7	10.07	10.6	10.5	13.55
Linear coefficient of	solid	2.91	1.75			
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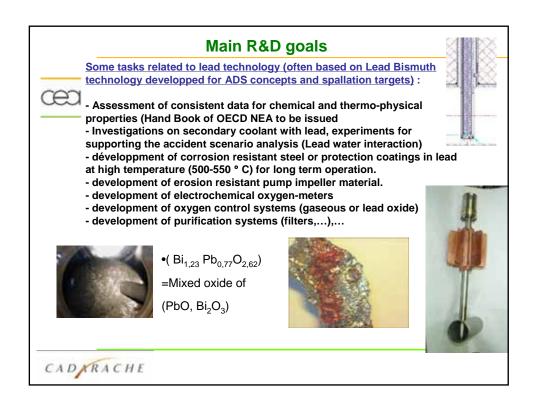
LME - lead/magnesium eutectic ** LBE - lead/bismuth eutectic

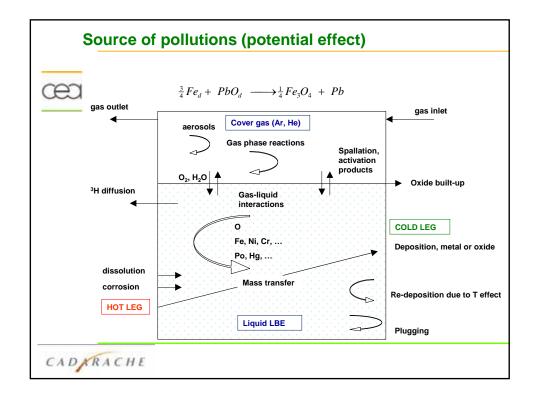




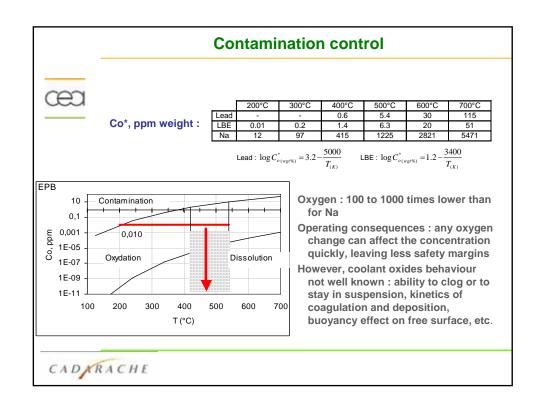


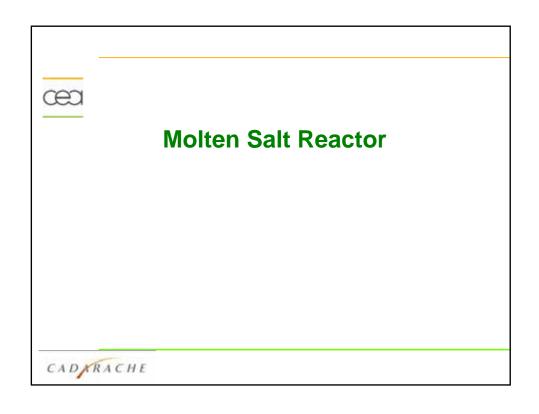


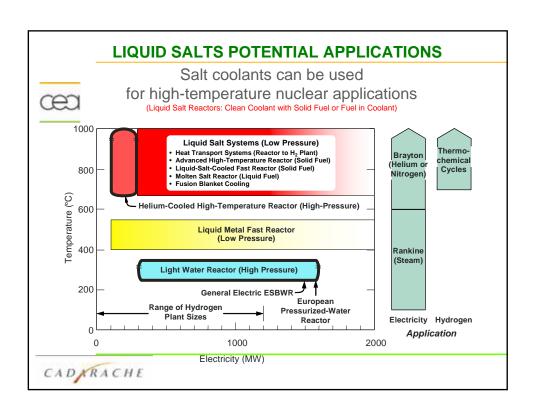












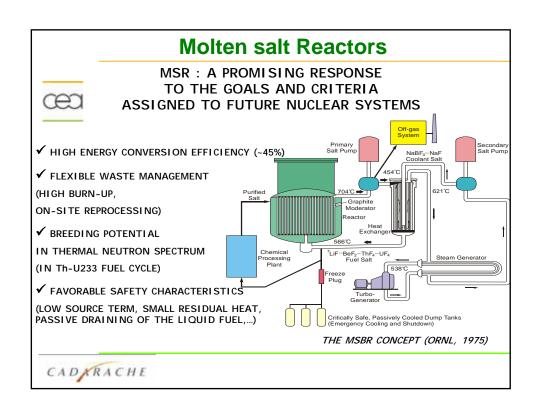
MSR SYSTEM IN GENERATION IV



MSR R&D SCOPE IN GENERATION IV

- 1. DEMONSTRATION OF FEASIBILITY AND EVALUATION OF PERFORMANCE OF TMSR FAMILY (TH-U233 BREEDER) ACCORDING TO GENERATION IV CRITERIA
- 2. EVALUATION OF PERFORMANCE OF MSR FOR WASTE REDUCTION (LWR SPENT FUEL)
- 3. EXPLORATION OF THE POTENTIAL OF LIQUID SALTS AS
 EFFICIENT COOLANTS FOR SOLID FUEL REACTORS, AS AN
 ALTERNATIVE TO SODIUM AND HELIUM:
 IN FAST REACTORS (LSFR)
 IN THERMAL SPECTRUM REACTORS (AHTR)
- 4. UTILIZATION OF LIQUID SALTS FOR HEAT TRANSPORT :
 SFR INTERMEDIATE LOOP
 COUPLING FLUID REACTOR-H2 UNIT

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Molten salt as intermediate coolant for SFR



Selection criteria:

Low melting point and liquid between 200 (or lower)-700 ° C Stable in a very large range of temperature

Good thermal properties (Cp, Pr number)

Compatibility with primary Na, water, CO₂, air,

Corrosion mastered

Relatively low pumping power

Low toxicity

Non flamable

Availability, low cost

Capability for structures to be cleaned and/or decontaminated

Low impact on decommissioning operations

Good wettability for US sensors (for In Service Inspection),...

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LIQUID SALTS POTENTIAL APPLICATIONS



Potential of liquid salts as coolants

		helium	CO ₂	water	sodium	FLiBe
		60 bar	60 bar	150 bar	1 bar	T >
	unit	500°C	500°C	300°C	500°C	450°C
ρ	kg/m ³	3.7	40.9	726	865	1940
c_p	kJ/kg.K	5.2	1.2	5.6	1.3	2.3
ρc_p	kJ/m ³ .K	19.4	48.6	4066	1125	4540
λ	W/m.K	0.29	0.06	0.56	80	1.0
μ	10 ⁻⁵ *Pa/s	3.8	3.3	9.0	23.3	563
Pr	-	0.67	0.66	0.90	0.004	13.2
relative merit*						
(ref. Na)	-	0.001	0.003	51	1	14

*merit factor = $c_p^{2,8} \rho^2 \mu^{-0,2}$

Good compatibility with water, air,...Non flamable, low toxicity, low cost,...

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Some other examples of salts:



Sel	Composition	Tfusion	Tmax	Density at 300 ° C	Used for solar plants
Hitec	7%NaNO ₃ 40%NaNO ₂ 53%KNO ₃	142 ° C	535 ° C	1,6	THEMIS (2,5MWe)
Drawsalt	46%- 60%NaNO ₃ 54%- 40%KNO ₃	220 ° C	600°	1,9	SOLAR 2 (10MWe)

➤ Salts at low melting point to be compared / criteria previously defined.

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